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### Structure of $[\text{As}(\text{C}_6\text{H}_5)_4]_2\text{-}[\text{Sn}\{\text{Se}_2\text{C}=\text{C}(\text{CN})_2\}_3]^{2-}$

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#### Abstract

In the complex anion  $[\text{Sn}\{\text{Se}_2\text{C}=\text{C}(\text{CN})_2\}_3]^{2-}$  in bis(tetraphenylarsonium) tris(2,2-diselenido-1,1-ethylenedicarbonitrile-*Se, Se'*)stannate(IV), the Sn atom is at the center of a distorted octahedron formed of six Se atoms of the chelate ligands with an average Sn—Se bond length of 2.661 (3) Å. The bis(selenium) chelates show only small deviations from planarity and the average Se—C distance is 1.88 (1) Å.

#### Comment

Up to now only a few tris(chelate) complexes with unsaturated 1,1-diselenolato ligands have been structurally characterized. For these transition-metal complexes two different coordination geometries have been found. Whereas the central ion,  $\text{Ni}^{\text{IV}}$  in  $[\text{Ni}\{\text{Se}_2\text{CN}(\text{C}_4\text{H}_9)_2\}_3]\text{Br}$  (Beurskens & Cras, 1971) and  $[\text{As}(\text{C}_6\text{H}_5)_4][\text{Ni}\{\text{Se}_2\text{C}=\text{C}(\text{CN})_2\}_3]$  (Kaiser, Dietzsch, Richter, Golič & Šiftar, 1980) has a distorted octahedral arrangement, the complex  $[\text{Fe}\{\text{Se}_2\text{CN}(\text{CH}_2\text{Ph})_2\}_3]\text{BF}_4$  has  $D_3$  macrosymmetry with a coordination polyhedron intermediate between the idealized trigonal prismatic and trigonal antiprismatic geometries (Deplano, Trogu, Bigoli, Leporati, Pellinghelli, Perry, Saxton & Wilson, 1983). Here we describe the X-ray crystal structure

analysis of the main-group metal tris(chelate) complex bis(tetraphenylarsonium) tris(2,2-diselenido-1,1-ethylenedicarbonitrile-*Se, Se'*)stannate(IV).

The geometry of the anion (Fig. 1) is comparable with  $[\text{Ni}\{\text{Se}_2\text{C}=\text{C}(\text{CN})_2\}_3]^{2-}$ ; the six Se atoms define an irregular octahedron and the average distance to the central metal is 2.661 (3) Å. The coordination around the  $\text{Sn}^{\text{IV}}$  is distorted octahedral. The average projected twist angle of the two  $\text{Se}_3$  triangles (Fig. 2) of the  $\text{SnSe}_6$  polyhedron is  $43.2^\circ$ , intermediate between the  $60$  and  $0^\circ$  expected for octahedral and trigonal prismatic geometry, respectively.

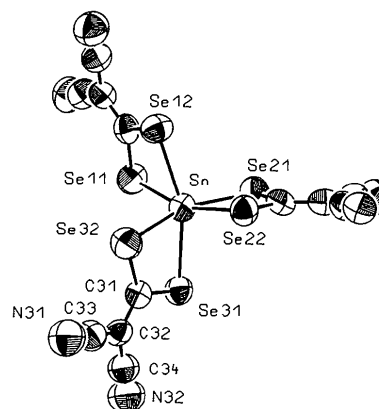


Fig. 1. Structure of the  $[\text{Sn}\{\text{Se}_2\text{C}=\text{C}(\text{CN})_2\}_3]^{2-}$  complex anion.

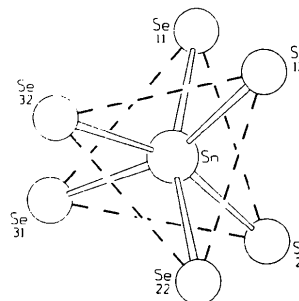
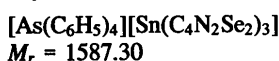


Fig. 2.  $\text{SnSe}_6$  core with the two  $\text{Se}_3$  triangles.

The average Se—Sn—*Se'* angle involving two different chelate ligands is  $160.4(1)^\circ$  ( $180^\circ$  in the perfect octahedron) while the intra-chelate angles Se—Sn—Se are in the narrow range  $73.7(1)$ – $74.0(1)^\circ$  and reflect the rigid chelate geometry with Se—C—Se of about  $116^\circ$ .

The three selenolate ligands are nearly planar ( $\Delta_{\text{max}} = 0.3$  Å) and the system of bond lengths and angles compares well with values expected for covalent coordinated diselenolates (Dietzsch, Franke, Hoyer, Grubb & Hummel, 1991).

**Experimental***Crystal data*

$$M_r = 1587.30$$

Triclinic

 $P\bar{1}$ 

$$a = 14.581(9) \text{ \AA}$$

$$b = 14.62(1) \text{ \AA}$$

$$c = 16.88(1) \text{ \AA}$$

$$\alpha = 102.40(2)^\circ$$

$$\beta = 105.45(2)^\circ$$

$$\gamma = 113.53(2)^\circ$$

$$V = 2963(3) \text{ \AA}^3$$

$$Z = 2$$

$$D_x = 1.779 \text{ Mg m}^{-3}$$

Mo  $K\alpha$  radiation

$$\lambda = 0.7099 \text{ \AA}$$

*Data collection*

Huber diffractometer

Lehmann-Larsen profile

analysis;  $\theta$  scan width  
( $1.0 + 0.20 \tan \theta$ ) $^\circ$ ;  $\theta$  scan  
rate  $0.8^\circ \text{ min}^{-1}$ 

Absorption correction:

empirical

$$T_{\min} = 0.836, T_{\max} = 1.150$$

5505 measured reflections

*Refinement*Refinement on  $F$ 

$$\text{Final } R = 0.062$$

$$wR = 0.055$$

$$S = 1.712$$

4572 reflections

455 parameters

H-atom parameters not refined

$$w = 1/[\sigma^2(F) + 0.0001F^2]$$

Data collection: *PWCOM* (Gomm, 1992). Cell refinement: *PWCOM*. Data reduction: *Xtal3.0 ADDREF*, *SORTRF* (Hall & Stewart, 1990). Program(s) used to solve structure: *SHELXS86* (Sheldrick, 1986). Program(s) used to refine structure: *SHELX76* (Sheldrick, 1976). Molecular graphics: *ORTEP* (Johnson, 1965); *SCHAKAL* (Keller, 1988). Software used to prepare material for publication: *Xtal3.0 BONDLA*, *CIFLO*.

Table 1. Fractional atomic coordinates and equivalent isotropic thermal parameters ( $\text{\AA}^2$ ) for the complex anion

Atomic coordinates for the cation have been deposited (see footnote).

$$U_{\text{eq}} = \frac{1}{3} \sum_i \sum_j U_{ij} a_i^* a_j^* \cdot \mathbf{a}_i \cdot \mathbf{a}_j$$

	x	y	z	$U_{\text{eq}}$
Sn1	0.2107 (1)	0.5765 (1)	0.2105 (1)	0.0574 (5)
Se11	0.2807 (1)	0.7200 (1)	0.1378 (1)	0.0670 (7)
Se12	0.4082 (1)	0.7284 (1)	0.3273 (1)	0.0670 (7)
C11	0.4063 (9)	0.7932 (8)	0.2424 (7)	0.057 (2)
C12	0.4912 (9)	0.8873 (9)	0.2555 (8)	0.060 (2)
C13	0.585 (1)	0.9453 (9)	0.3350 (9)	0.063 (2)
C14	0.4933 (9)	0.9305 (9)	0.1890 (9)	0.067 (2)
N11	0.6582 (9)	0.9921 (9)	0.4008 (8)	0.098 (2)
N12	0.4908 (8)	0.9619 (8)	0.1325 (8)	0.085 (2)

Se21	0.2504 (1)	0.4373 (1)	0.1159 (1)	0.0694 (8)
Se22	0.2057 (1)	0.4383 (1)	0.2919 (1)	0.0634 (7)
C21	0.2278 (8)	0.3687 (9)	0.1970 (8)	0.056 (2)
C22	0.2294 (9)	0.275 (1)	0.1879 (8)	0.064 (2)
C23	0.237 (1)	0.217 (1)	0.1130 (9)	0.075 (2)
C24	0.2206 (9)	0.227 (1)	0.2514 (9)	0.074 (2)
N21	0.2418 (9)	0.1732 (9)	0.0508 (8)	0.097 (2)
N22	0.2152 (8)	0.1901 (8)	0.3053 (8)	0.085 (2)
Se31	0.0009 (1)	0.4826 (1)	0.1063 (1)	0.0626 (7)
Se32	0.1185 (1)	0.6544 (1)	0.3031 (1)	0.0766 (8)
C31	-0.0111 (8)	0.5649 (8)	0.2015 (7)	0.049 (2)
C32	-0.1080 (9)	0.5611 (8)	0.1956 (8)	0.054 (2)
C33	-0.1169 (9)	0.618 (1)	0.2678 (9)	0.074 (2)
C34	-0.199 (1)	0.506 (1)	0.1148 (9)	0.080 (2)
N31	-0.1236 (9)	0.6669 (9)	0.3284 (8)	0.106 (2)
N32	-0.2698 (9)	0.4648 (9)	0.0482 (8)	0.097 (2)

Table 2. Geometric parameters ( $\text{\AA}$ ,  $^\circ$ )

Sn1—Se11	2.648 (3)	Se22—C21	1.88 (1)
Sn1—Se21	2.658 (3)	C21—C22	1.36 (2)
Sn1—Se31	2.661 (2)	C22—C24	1.41 (2)
Sn1—Se12	2.665 (2)	C22—C23	1.42 (2)
Sn1—Se22	2.666 (3)	C23—N21	1.14 (2)
Sn1—Se32	2.669 (3)	C24—N22	1.16 (2)
Se11—C11	1.875 (9)	Se31—C31	1.88 (1)
Se12—C11	1.88 (1)	Se32—C31	1.887 (8)
C11—C12	1.36 (1)	C31—C32	1.37 (2)
C12—C14	1.40 (2)	C32—C33	1.39 (2)
C12—C13	1.41 (1)	C32—C34	1.41 (2)
C13—N11	1.14 (1)	C33—N31	1.17 (2)
C14—N12	1.14 (2)	C34—N32	1.13 (2)
Se21—C21	1.89 (1)		
Se11—Sn1—Se21	94.64 (9)	C14—C12—C13	116 (1)
Se11—Sn1—Se31	96.63 (8)	N11—C13—C12	178 (2)
Se11—Sn1—Se12	73.83 (8)	N12—C14—C12	177.1 (9)
Se11—Sn1—Se22	160.77 (8)	Se22—C21—Se21	116.6 (7)
Se11—Sn1—Se32	102.30 (9)	C22—C21—Se22	122 (1)
Se21—Sn1—Se31	94.91 (7)	C22—C21—Se21	121 (1)
Se21—Sn1—Se12	103.32 (9)	C21—C22—C24	122 (1)
Se21—Sn1—Se22	74.03 (8)	C21—C22—C23	123 (1)
Se21—Sn1—Se32	160.48 (7)	C24—C22—C23	115 (1)
Se31—Sn1—Se12	159.93 (9)	N21—C23—C22	176 (2)
Se31—Sn1—Se22	99.74 (7)	N22—C24—C22	178 (1)
Se31—Sn1—Se32	73.74 (8)	Se31—C31—Se32	116.0 (7)
Se12—Sn1—Se22	93.37 (8)	C32—C31—Se31	121.4 (7)
Se12—Sn1—Se32	90.79 (8)	C32—C31—Se32	122.6 (9)
Se22—Sn1—Se32	91.96 (9)	C31—C32—C33	120.7 (9)
Se11—C11—Se12	116.4 (5)	C31—C32—C34	121 (1)
C12—C11—Se11	122 (1)	C33—C32—C34	118 (1)
C12—C11—Se12	122.0 (8)	N31—C33—C32	179 (1)
C11—C12—C14	121.7 (9)	N32—C34—C32	177 (2)
C11—C12—C13	122 (1)		

Lists of structure factors, anisotropic thermal parameters, coordinates of the cation, bond distances, bond angles, torsion angles and mean-planes data, have been deposited with the British Library Document Supply Centre as Supplementary Publication No. SUP 55905 (34 pp.). Copies may be obtained through The Technical Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England. [CIF reference: SH1016]

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### Structure of Bis- $\mu$ -[(trimethylsilyl)methanolato-*O*]-bis(trimethylsilylmethyl)gallium] and Bis- $\mu$ -[(trimethylsilyl)methanolato-*O*]-bis(trimethylsilylmethyl)indium]

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#### Abstract

Two new compounds, (I) bis- $\mu$ -(trimethylsilyl)methanolato-*O*-bis[bis(trimethylsilylmethyl)gallium] and (II) bis- $\mu$ -(trimethylsilyl)methanolato-*O*-bis[bis(trimethylsilylmethyl)indium], have been obtained with dimeric structures, with planar four-membered  $\text{Ga}_2\text{O}_2$  and  $\text{In}_2\text{O}_2$  rings, respectively.

#### Comment

The asymmetric unit of (I) consists of two independent molecules each with an inversion centre in the middle of the four-membered ring. As these two molecules are almost identical, only one is shown in Fig. 1 (weighted r.m.s deviation 0.17 Å). Owing to the inversion centre the four-membered ring is exactly planar. On average, the endocyclic angles are O—Ga—O 81.6 (1) and Ga—O—Ga 98.4 (1)°. The exocyclic angles are increased by the steric

demand of the bulky  $\text{CH}_2\text{Si}(\text{CH}_3)_3$  groups [C—Ga—C 126.8 (2)°]. The observed average bond lengths Ga—O 1.967 (4) and Ga—C 1.966 (4) Å agree well with values found in the analogous compound  $[(\text{C}_5\text{H}_5)_2\text{GaOEt}]_2$  (Cowley, Mehrota, Atwood & Hunter, 1985).

In contrast to (I), compound (II) does not possess any crystallographic symmetry. The four-membered  $\text{In}_2\text{O}_2$  ring is almost planar (mean deviation from least-squares plane 0.06 Å). The mean endocyclic angles are O—In—O 79.3 (1) and In—O—In 100.3 (1)°. As in (I), the exocyclic angles on the In atoms are increased by the steric demand of the  $\text{CH}_2\text{Si}(\text{CH}_3)_3$  groups. The different values for each In atom are caused by the different arrangement of the  $\text{CH}_2\text{Si}(\text{CH}_3)_3$  groups [C5—In1—C6 130.9 (1) and C3—In2—C4 144.2 (1)°]. As in (I), no unusual bond lengths were found. The mean bond lengths In—O 2.173 (2) and In—C 2.152 (4) Å are in good agreement with values found in  $[(\text{tBu})\text{InOEt}]_2$  (Bradley, Frigo, Hursthouse & Hussain, 1988).

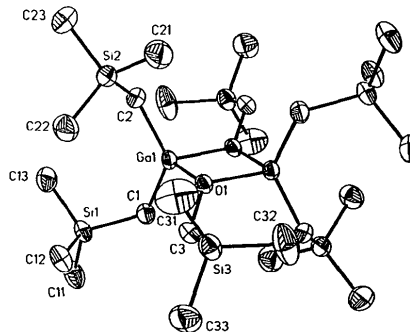


Fig. 1. Structure of one molecule of (I) showing 50% probability displacement ellipsoids. The H atoms are omitted for clarity.

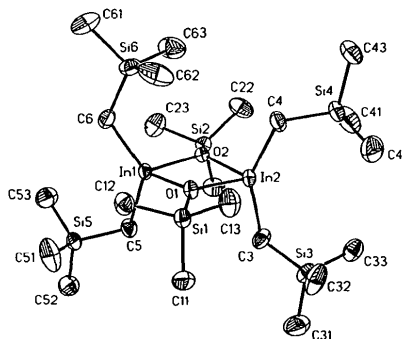


Fig. 2. Structure of (II) showing 50% probability displacement ellipsoids. The H atoms are omitted for clarity.

#### Experimental

##### Compound (I)

###### Crystal data

$[\text{Ga}_2(\text{C}_4\text{H}_{11}\text{OSi})_2(\text{C}_4\text{H}_{11}\text{Si})_4]$   
 $M_r = 694.75$

$D_x = 1.131 \text{ Mg m}^{-3}$   
Mo  $K\alpha$  radiation